CE 329, Fall 2015

Assignment 29

Problem Statement

Irreversible reaction (1) takes place in the liquid phase with a rate that is second order in the reactant. Suppose an isothermal recycle PFR was going to be used for this reaction and that it would operate at a temperature where the rate coefficient has a value of 193 ft³ lbmol⁻¹ h⁻¹. The stream to be processed has a flow rate of 21.2 ft³ h⁻¹ and contains 0.075 lbmol of A per ft³. The PFR is 10 ft long and has a diameter of 4.4 inches. What will the overall conversion equal if the recycle ratio is zero? What will the overall conversion equal if the recycle ratio is 0.5?

$$A \rightarrow R$$
 (1)